

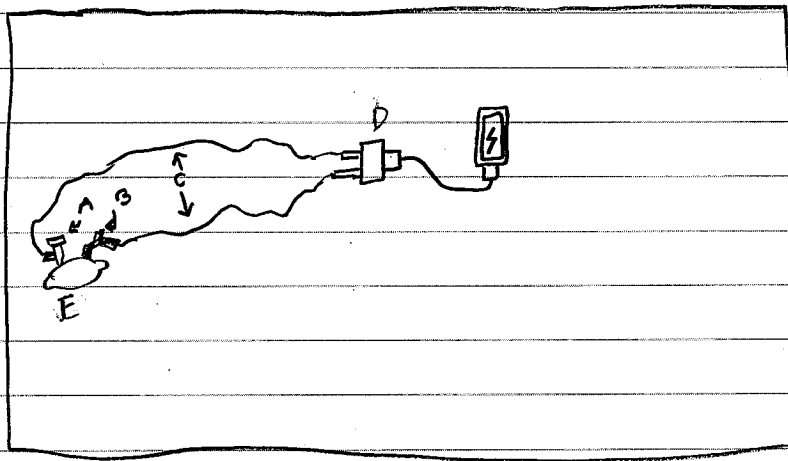
# Topic = Charging a phone using lemons

1 Lemon produces 0.9-1.0 volts

Quantity of amps tells you how much electricity is being drawn through the power cable - Volts also must be correct for a product to function

1000 miliamps = 1 amp

Hypothesis (Plan if works)



Find Wattage  
 = # Volts  
 x  
 # Amps  
 =  
 # Watts

A) Zinc covered nails (-)

B) Medium gauge copper wire (+)

C) Alligator clips

D) Outlet + USB cable

E) Lemons

USB 2.0

5V x 0.5A

USB 3.2

2.5W

USBC

5V x 3A

USB 3.0



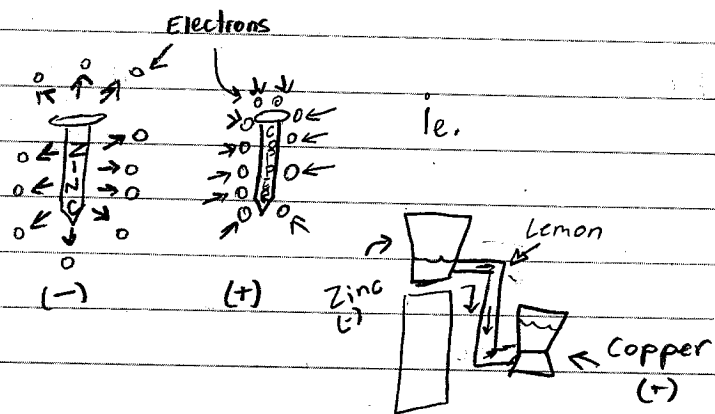
(No wrong way of putting it in)

15W

5V x 0.9A

4.5W

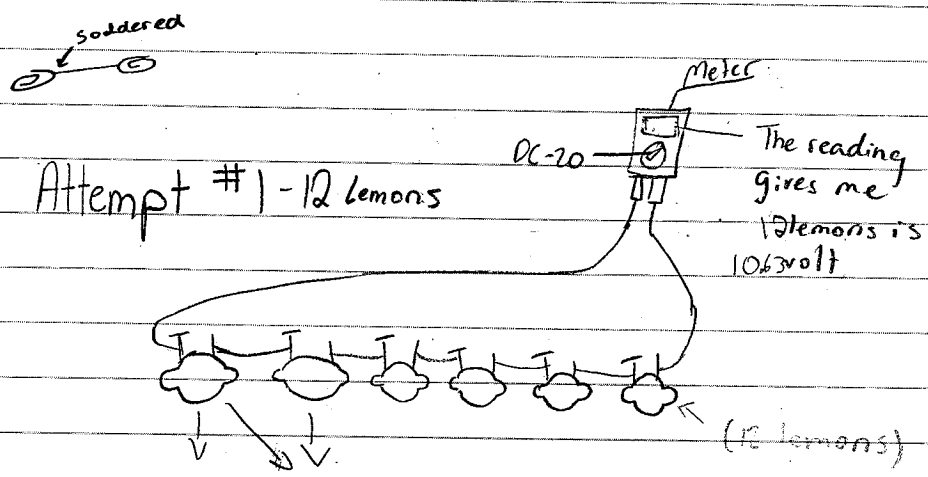
Has the potential to transfer a lot more power



Cathodes usually made of copper  
 Anodes usually made of zinc

30 lemons produce 26 volts (18 Batteries)

New Procedure:



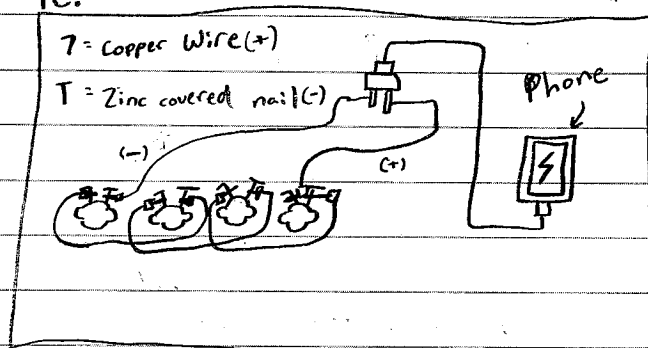
Attempt #1 - 12 lemons

	0.93	1.9				
	DC	AC		DC		DC AC
1+	0.93	N/A	6	4.93	11	9.56 N/A
+2	1.89	N/A	7	5.87	12	10.47 0.006
3	2.84	N/A	8	6.78		
4	3.34	N/A	9	7.74		
5	4.02	N/A	10	8.68		

Manufacturers want to make more powerful devices that have higher energy requirements

Lemon circuit

ie:



To find how much lemons needed to charge

example:

$$5 \div 0.9 = 5.55$$

(Volts) (Lemon Charge) (# of lemons needed)

Round to nearest whole so 6 lemons

12/26/23

Established standard for charging mobile phones is 5 volts DC from a USB port or charger + 2.4 amps

Quantity of lemons needed to charge phone (Hypothesis)

$$5 \div 0.9 = 5.55 = 6 \text{ lemons needed}$$

# AC $\rightarrow$ Alternate Current - Crucial in Phone charging

Conclusion: When it was revealed later that the charger is a AC to DC converter, the DC somehow didn't go through. To possibly do this, we need to have <sup>at least</sup> a thousand lemons to match the AC to DC converter.

